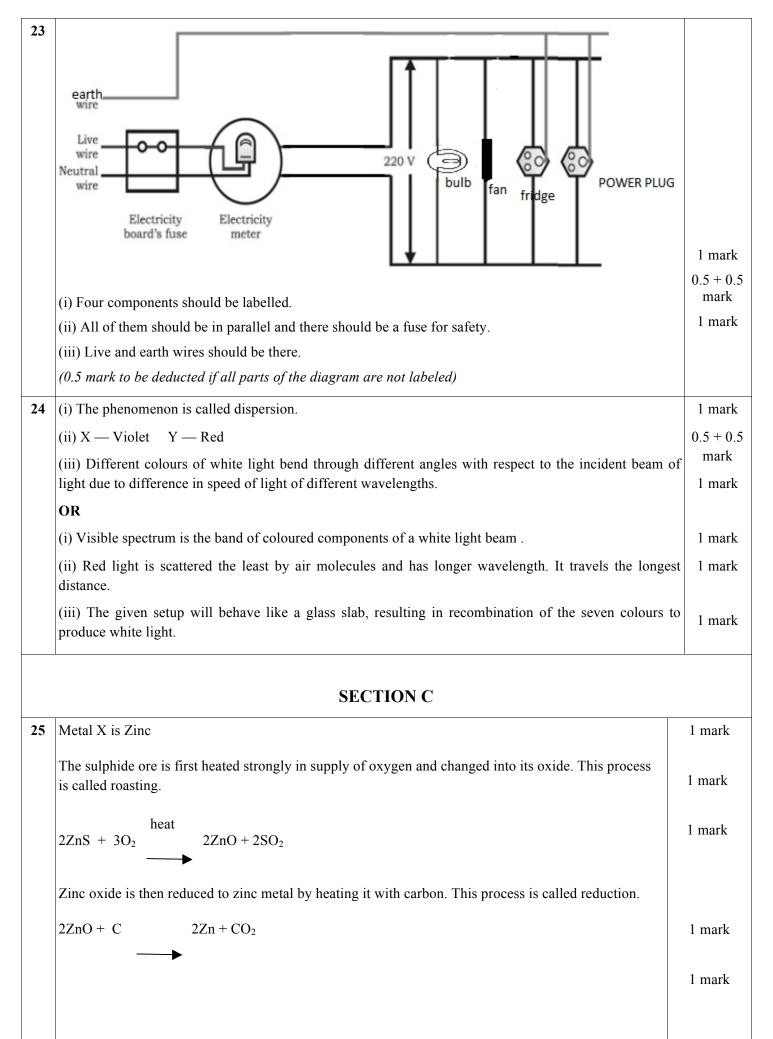
Physics Master Academy Only Teaching Noting ElseCLASS X SCIENCE: 086

Marking Scheme for Sample Question Paper 2019-20

SECTION A			
1	The property of self-linking of atoms of an element through covalent bonds in order to form straight chain, branched chains or cyclic chains of different sizes is called catenation.	1 mark	
2	The valency of an element first increases and then decreases across a period.	1 mark	
3(a)	The principle behind electric generator is Electromagnetic Induction- the phenomenon of producing current in a coil by changing the magnetic field associated with it.	1 mark	
3(b)	The polarity of the output alternating current changes every 1/100 seconds. Alternately:	1 mark	
	In 1 second the output (AC) completes 50 cycles.		
3(c)	The suitability of Muppandal as a site for wind farms stems from its geographical location as it has access to the seasonal monsoon winds.	1 mark	
3(d)	City A	0.5 mark	
	It is more suitable for a wind-farm as there is consistently high wind-speed in that city throughout the year.	0.5 mark	
4(a)	Diabetes	1 mark	
4(b)	Insulin	1 mark	
4(c)	iv) low sugar high fibre diet	1 mark	
4(d)	i) 180mg/dL	1 mark	
5	ii) pupils take time to adjust	1 mark	
	OR		
	ii) refraction		
6	$ii) = 40 \Omega$.	1 mark	
	V=IR, V = 4V, I = 100 mA = 0.1 A		
	Hence $R = V/I = 4/0.1 \ \Omega = 40 \ \Omega$.		
7	i) volt-ampere	1 mark	
	Power = Voltage x Current.		
8	iv) Human faecal matter	1 mark	
	OR		
	iv) The Industralist		
9	iii) Carbon monoxide	1 mark	
10	iv) Decomposition of calcium carbonate to form quick lime and carbon dioxide.	1 mark	
11	i) Na ₂ CO ₃	1 mark	

12	iii) C	1mark
	OR	
	iii) Q and R	
13	i) Both assertion and reason are true and reason is the correct explanation of assertion.	1 mark
14	iv) Assertion is false but reason is true.	1 mark
	SECTION B	
15	(i) It turns yellow due to formation of lead oxide and Reddish brown fumes evolve.	0.5 + 0.5
	(ii) Thermal decomposition reaction.	mark
	heat	
	(iii) $2Pb(NO_3)_2 \longrightarrow 2PbO + 4NO_2 + O_2$	1 mark
		1 mark
16	(i) Sodium bicarbonate/Sodium hydrogencarbonate/ baking soda and its formula is NaHCO ₃	0.5 + 0.5 mark
	(ii) $2NaHCO_3 \longrightarrow Na_2CO_3 + CO_2 + H_2O$	1 mark
	(iii) It is used in fire extinguisher and for baking. (any one)	1 mark
	OR	
	(i) Add 10 mL of concentrated sulphuric acid slowly to 90 mL of water with constant stirring.	1 mark
	Dilution of acid is a highly exothermic process. If water is added to concentrated sulphuric acid, heat	1 mark
	generated causes the mixture to splash leading to burns and the glass container can break.	1 mark
	(ii) Decreases per unit volume.	
17	Electronic configuration of X: 2,8,2, Y: 2,8,6	0.5 + 0.5
	Both X and Y belong to 3rd period.	mark
	Ionic bond will be formed.	0.5 mark
	Reason: X will lose 2 electrons and Y will gain 2 electrons to complete their octet and become stable.	1 mark
	Formula is XY	0.5 mark
18	A food chain showing Ist tropic level (¹ / ₂ mark), II nd tropic level (¹ / ₂ mark), III rd tropic level	2 mark
	$(\frac{1}{2} \text{ mark})$ and IV th tropic level ($\frac{1}{2} \text{ mark}$).	
	A flow chart or a diagrammtic representation showing all the four tropic levels would also be accepted	
	According to the 10% law, the amount of energy available will not be sufficient for the survival of the organism in the 5th trophic level.	1 mark
	OR	
	• Large jar filled with water, oxygen, food and aquatic plants and animals.	
	Oxygen/oxygen pump.	
	• Fish food.	
	• Aquatic plants/Producers provide O ₂ during photosynthesis.	
	 Aquatic animals/Consumers release CO₂ for the process of photosynthesis. 	3 mark
	 Decomposers are also important for natural cleaning of the aquarium. 	
	(0.5 mark for each point)	

19	Glucose (6-Carbon molecule) In Cytoplasm Merce of (Yeast) Pyruvate (3-Carbon molecule) + Energy Presence of O ₂ (Mitochondria) CO ₂ + H ₂ O + Energy CO ₂ + H ₂ O + Energy	1+1+1 mark
20	Let purple trait be represented by: PP White trait be : pp parental PP X pp \downarrow F_1 Pp X Pp (Selfing) F_2	0.5 + 1 + 0.5 mark
	Gametes P p	
	P PP Pp	
	p Pp pp Visible characters of F1 progeny all flowers are purple coloured and in F2 progenies 3 are purple coloured and 1 is white coloured flower	0.5 + 0.5 mark
21	When growing plant shoot tip detects light a hormone called auxin is synthesised in the shoot tip which is sensitive to light.Auxin diffuses towards the shady side of the stem.It stimulates the growth of cells on the shady side of the plant which causes bending of the plant to the other side.	0.5mark 1 mark
	This gives the appearance that the stem of the plant bends in the direction of light.	0.5 mark
22	(i) Range of distance should be 0 cm to < 12 cm.	0.5 mark
	(ii) The image will larger than the object.	1.5 mark
	(0.5 mark to be deducted if no arrows marked or wrongly marked arrows)	
		1



	OR	
	(i) As it reacts with both acids as well as bases to form salts.	1 mark
	 (ii) Iron being more reactive than copper displaces copper from copper sulphate to form green ferrous sulphate solution. 	1 mark
	 (iii) Nitric acid is a strong oxidising agent. Hydrogen gas produced gets oxidised to H₂O. (iv) Calcium is a very reactive metal. It reacts with the chemicals in surroundings and occurs in combined state. 	1 mark
	 (v) Sodium and potassium are highly reactive metals and react vigorously with oxygen in air and may even catch fire. They do not react with kerosene. 	1 mark 1 mark
26	(i) D is a saturated hydrocarbon	0.5 mark
	(ii) B is an organic acid. Structural formula $ \begin{array}{c} H \\ H \\ H \\ H \\ O \\ H \\ O \\ H \end{array} $	0.5 mark
	Ч ́О-Н	0.5 mark
	(iii) C is the compound.	0.5 mark
	It acts as a dehydrating agent and removes a water molecule from ethanol.	0.5 mark
	hot conc. H ₂ SO ₄	1 mark
	C_2H_5OH $C_2H_4 + H_2O$	
	(iv)	
	conc H_2SO_4	1 mark
	$CH_3COOH + C_2H_5OH \longrightarrow CH_3COOC_2H_5 + H_2O$	0.5 mark
	Major product is Ester and it is used in making perfumes / flavouring agents.	
27	 (i) Oxygenated : B/D/F [B= left ventricle/D=aorta/F=left auricle/pulmonary vein] Deoxygenated: A/C/E [A= right ventricle/C= pulmonary artery/E=right auricle/vena cava] (any two) 	0.5 + 0.5 + 0.5 +0.5 mark



